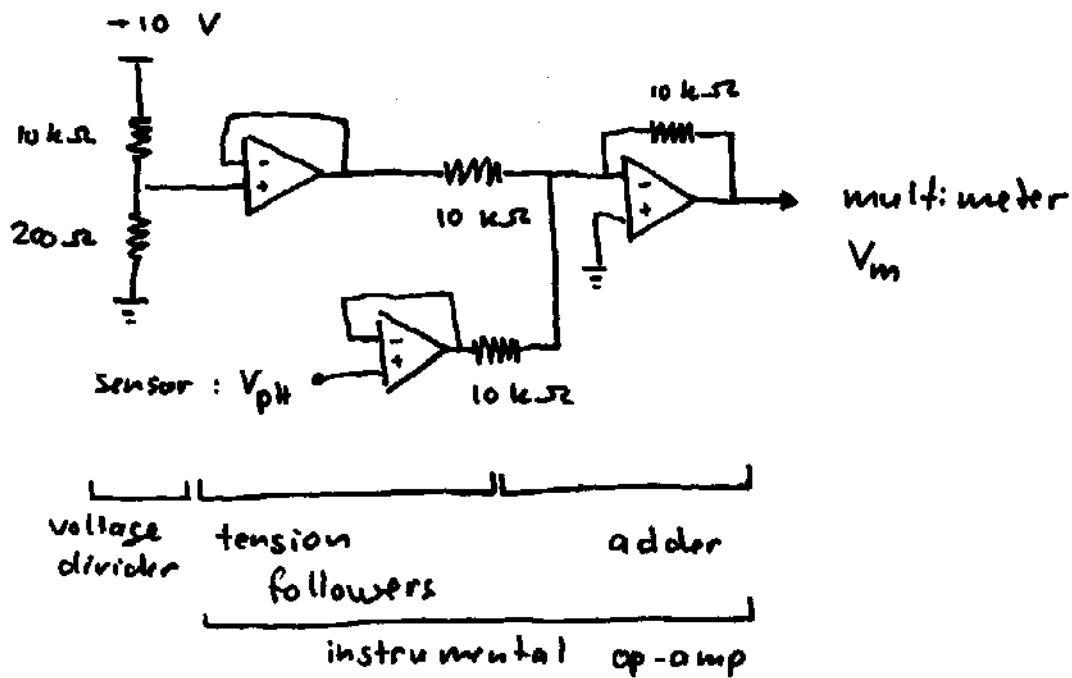


- ③ a) The signal at a pH of 5 is equal to +200 mV. To get highest resolution we have to subtract this voltage.



b) $\Delta N_{H^+} = \frac{\Delta V_m}{\frac{dV_m}{dN_{H^+}}}$, $\Delta V_m = 0,01 \text{ mV}$

$$\frac{dV_m}{dN} = \frac{dV_m}{d\text{pH}} \cdot \frac{d\text{pH}}{dN} = \frac{dV_m}{dV_{\text{pH}}} \cdot \frac{dV_{\text{pH}}}{d\text{pH}} \cdot \frac{d\text{pH}}{d[N]} \cdot \frac{d[N]}{dN}$$

$$\left. \begin{aligned} \frac{dV_m}{dV_{\text{pH}}} &= 1 \quad (\text{see circuit above}) \\ \frac{dV_{\text{pH}}}{d\text{pH}} &= 100 \text{ mV} \\ \frac{d\text{pH}}{dN} &= \frac{1}{\ln(10)} \cdot \frac{1}{[N_{H^+}]} \\ \frac{d[N]}{dN_{H^+}} &= \frac{1}{3l} \end{aligned} \right\} \frac{dV_m}{dN_{H^+}} = \frac{100 \text{ mV}}{\ln(10)} \cdot \frac{1}{[N_{H^+}]} \cdot \frac{1}{3l}$$

$$\text{pH} = 5 \Rightarrow [N_{H^+}] = 10^9 / \text{liter}$$

$$\frac{dV_m}{dN_{H^+}} = 1.45 \cdot 10^{-11} \text{ V / atom } H^+$$

$$\Rightarrow \Delta N_{H^+} = \frac{10 \text{ } \mu\text{V}}{1.45 \cdot 10^{-11} \text{ V}} \quad \text{atoms } H^+ = 6.9 \times 10^5 \text{ atoms}$$

④ see lecture notes

- thermo couple
 - PT 100
 - thermistor
 - diode
 - mechanical sensor
 - spectrum analyzer
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⑤ see lecture notes

